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(Z)-1-Benzoyl-4-*tert*-butyl-1-methylcyclohexane and (Z)-4-*tert*-Butyl-1-methyl-1thiobenzoylcyclohexane[†]

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Abstract

Both title molecules, $C_{18}H_{26}O$ and $C_{18}H_{26}S$, contain sixmembered cyclohexane rings with chair conformations, equatorial 1-methyl and 4-*tert*-butyl substituents, and axial 1-benzoyl or 1-thiobenzoyl substituents, respectively. The substituent carbonyl and thiocarbonyl groups are oriented with the O and S atoms close to axial H atoms in the cyclohexane rings, thus accounting for the observed photochemical reactivity.

Comment

The materials were synthesized as part of a study of asymmetric induction in the photochemistry of ketones (Leibovitch et al., 1996; Leibovitch, 1997). Each molecule of (I) and (II) contains a six-membered cyclohexane ring with a chair conformation, with torsion angles $\pm 54.2-57.9(3)$ and $51.3-57.2(2)^{\circ}$, respectively. The 1-methyl and 4-tert-butyl substituents are in equatorial positions, and the 1-benzoyl and 1-thiobenzoyl substituents are in axial sites. The phenyl groups are rotated considerably out of the planes of the carbonyl and thiocarbonyl groups; O-C-C-C torsion angles are 136.3 (3) and -41.6 (4)°, and S-C-C-C torsion angles are 128.9(2) and $-48.3(2)^{\circ}$, for the benzoyl, (I), and thiobenzoyl compound, (II), respectively. Bond lengths and angles are normal; C=O 1.220(3) and C=S 1.622(2) Å.



[†] Alternative names: *cis*-[4-(1,1-dimethylethyl)-1-methylcyclohexyl]phenylmethanone and *cis*-[4-(1,1-dimethylethyl)-1-methylcyclohexyl]phenylmethanethione.

The carbonyl and thiocarbonyl groups are oriented approximately normal to the C4...Cl axes of the cyclohexane rings; O/S—C7—C1—C6 torsion angles are 26.5 (3) and 35.5 (2)° in the two compounds. This brings the O and S atoms reasonably close to the axial H atoms on C5, with O...H 2.69 and S...H 3.03 Å, distances which are suitable for H-atom abstraction in the photochemical reactions (Leibovitch *et al.*, 1996) (the more distant axial H atoms on C3 are at 3.57 and 4.05 Å).



Fig. 1. Views of the two title molecules with 33% probability displacement ellipsoids.

Experimental

The title compounds were synthesized according to the methods of Leibovitch *et al.* (1996) and Leibovitch (1997).

Compound (I)

Crystal data

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Cu K\alpha radiation
C18H26O
                                      \lambda = 1.5418 \text{ Å}
M_r = 258.40
Orthorhombic
                                      Cell parameters from 24
                                         reflections
Pna2_1
                                      \theta=12.7{-}24.0^\circ
a = 20.120(4) Å
                                      \mu = 0.480 \text{ mm}^{-1}
b = 13.075(4) Å
                                      T = 294 \text{ K}
c = 6.149(4) Å
                                      Needle
V = 1618(1) \text{ Å}^3
Z = 4
                                      0.30 \times 0.10 \times 0.10 mm
D_x = 1.061 \text{ Mg m}^{-3}
                                      Colourless
D_m not measured
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Data collection

Rigaku AFC-6S diffractom-	1052 reflections with
eter	$I > 3\sigma(I)$
ω -2 θ scans	$\theta_{\rm max} = 77.64^{\circ}$
Absorption correction:	$h = -25 \rightarrow 0$
ψ scans (North, Phillips	$k = 0 \rightarrow 16$
& Mathews, 1968)	$l = 0 \rightarrow 7$
$T_{\rm min} = 0.962, T_{\rm max} = 1.000$	3 standard reflections
1814 measured reflections	every 200 reflections
1814 independent reflections	intensity decay: 29%

Refinement

Refinement on F^2 R(F) = 0.037 $wR(F^2) = 0.110$ S = 1.121814 reflections 172 parameters H atoms not refined $w = 1/[\sigma^2(F_o^2) + 0.00063(F_o^2)^2]$ $(\Delta/\sigma)_{max} = 0.0006$

Compound (II)

Crystal data

 $C_{18}H_{26}S$ $M_r = 274.46$ Orthorhombic *Pbca* a = 12.853 (1) Å b = 32.252 (3) Å c = 7.945 (1) Å $V = 3293.5 (6) Å^3$ Z = 8 $D_x = 1.107 Mg m^{-3}$ D_m not measured

Data collection

Rigaku AFC-6S diffractometer $\omega - 2\theta$ scans ψ scans (North, Phillips b) the & Mathews, 1968) $T_{min} = 0.699, T_{max} = 0.855$ 3492 measured reflections 3492 independent reflections *Refinement*

Absorption correction:

Refinement on F^2 R(F) = 0.047 $wR(F^2) = 0.119$ S = 2.36 3492 reflections 173 parameters H atoms not refined $w = 1/[\sigma^2(F_o^2) + 0.00005(F_o^2)^2]]$ $(\Delta/\sigma)_{max} = 0.016$

$h = 0 \rightarrow 16$ $k = 0 \rightarrow 40$ $l = 0 \rightarrow 10$ 3 standard reflections every 200 reflections

intensity decay: 12%

$$\begin{split} &\Delta\rho_{\text{max}}=0.32 \text{ e } \text{\AA}^{-3} \\ &\Delta\rho_{\text{min}}=-0.31 \text{ e } \text{\AA}^{-3} \\ &\text{Extinction correction:} \\ & \text{Zachariasen (1967)} \\ &\text{Extinction coefficient:} \\ & 2.2 (1) \times 10^{-6} \\ &\text{Scattering factors from} \\ & \text{International Tables for} \\ & Crystallography (Vol. C) \end{split}$$

The structures are not isomorphous, with space groups $Pna2_1$ (Z = 4) and Pbca (Z = 8), for the carbonyl and thiocarbonyl compounds, respectively.

For both compounds, data collection: MSC/AFC Diffractometer Control Software (Molecular Structure Corporation, 1992); cell refinement: MSC/AFC Diffractometer Control Software; data reduction: TEXSAN (Molecular Structure Corporation, 1995). Program(s) used to solve structures: SIR92 (Altomare, Cascarano, Giacovazzo & Guagliardi, 1993) for (I); SHELXS86 (Sheldrick, 1985) for (II). For both compounds, program(s) used to refine structures: TEXSAN; software used to prepare material for publication: TEXSAN.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: FG1298). Services for accessing these data are described at the back of the journal.

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ctom- $I > 3\sigma(I)$ $\theta_{max} = 77.68^{\circ}$

 $\Delta \rho_{\rm max} = 0.20 \ {\rm e} \ {\rm \AA}^{-3}$

 $\Delta \rho_{\rm min} = -0.22 \ {\rm e} \ {\rm \AA}^{-3}$

Extinction correction: Zachariasen (1967)

Extinction coefficient:

Scattering factors from

International Tables for

Crystallography (Vol. C)

 $2.7(6) \times 10^{-6}$

Cu $K\alpha$ radiation

Cell parameters from 23

 $0.12 \times 0.10 \times 0.10$ mm

 $\lambda = 1.5418 \text{ Å}$

reflections

 $\theta=42.3{-}50.4^\circ$

T = 294 K

Prism

Purple

 $\mu = 1.568 \text{ mm}^{-1}$